A NEW SYNTHESIS OF α, β -UNSATURATED KETONES

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Summary : α,β -unsaturated ketones have been obtained by thermolysis of δ -hydroxy β -keto triphenylphosphoranes prepared from the corresponding β -keto phosphoranes.

In a preceding paper (1), we have described a new extrusion reaction of triphenylphosphine oxide by thermolysis of the condensation product of a β -propiolactone and an ylide. (Scheme I)

SCHEME I



The scope of this practical synthesis of α , β -unsaturated ketones appears to be limited because of the relative inaccessibility of β -propiolactones.

In this communication, we show that the extrusion described above may be the key step in a new general method for the synthesis of α , β -unsaturated ketones. (Scheme II) The first step, (a), is a well known reaction : the acylation can be performed by the reaction of the ylide $\underline{1}$ with an acyl chloride (2), an acid anhydride (3), a thiol ester (4) or an aryl ester (5). The two latter reagents provide the keto-ylide $\underline{2}$ without the necessity of a transylidation.

The second step (b) has been previously described (6) with acetylmethylenetriphenylphosphorane (2, $R = R^1 = H$). We have found that this reaction is a general one and affords the compounds 3 in good yield whatever R, R^1 , R^2 and R^3 .

The pyrolysis is accomplished by heating at $150-170^{\circ}$ C under reduced pressure for 20-30 minutes. The ketone is condensed on a cooled trap. Examination of the ¹H-NMR spectra shows that the synthesis is stereoselective and affords only the α,β -unsaturated ketones possessing the E-configuration (95%).

SCHEME II



In principle, this synthesis could be envisioned to take place trough alkylation of the enclate anion followed by an alkydenation reaction. (Scheme III)

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Unfortunately, in this last scheme, when starting from disymmetric ketones, a mixture of different ethylenic ketones is obtained. In contrast, our method appears to be efficient because it permits a differentiation between α and α ' sites. (7)

The table below shows the possibilities offered by this new synthesis.

	Yield (%)	Eb (°C/mmHg)	Lit.data	3
<u>4a</u>	40	48-52/350	33-4/130	(9)
<u>4 b</u>	65	83-5/15	91-5/25	(10)
<u>4c</u>	72	80/5	-	
<u>4d</u>	45	70/6	68/9	(11)
<u>4e</u>	62	80/15	-	
<u>4f</u>	66	115-8/760	117-9/760	(12)
48	61	76-9/2	-	

Table α , β -unsaturated ketones 4

yields represent pure isolated products

As shown in the table, a range of substitutions are possible using this method. However, we observe that the pyrolysis of bulky alcohols $\underline{3}$ is sometimes difficult. Thus, the alcohol $\underline{3h}$ fails to give the corresponding α -ethylenic ketone $\underline{4h}$.

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